(54)	FRAGRANT LIQUID (CLEANING COMPOSITION				
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(57)	The present invention relates to fragrant liquid					
	cleaning compositions. More particularly it relates to					
	thickened fragrant liquid cleaning compositions with					
	longer lasting olfactory properties.					

Claim

1. A fragrant aqueous liquid cleaning composition comprising from 0.05-20% by weight of one or more surface-active detergents, from 0.1-1.5% by weight of one or more perfumes, and from 0.1-3% by weight of one or more perfumes in an aqueous medium, wherein the polymer is shear thinning, having a viscosity, at 25°C in deionised water at a concentration of 3% by weight or less, of between 0.3 and 0.006 Pa.s at a shear rate of 2000 sec⁻¹, of between 0.7 and 0.03 Pa.s at a shear rate of 30 sec⁻¹, and of between 0.6 and 50 Pa.s at a shear rate of 0.5 sec⁻¹.

Form 10

PATENTS ACT 1962

COMPLETE SPECIFICATION

(ORIGINAL)

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Name of Applicant:	TO BE COMPLETED BY APPLICANT	
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Countries Specification	for the invention entitled: FRAGRANT LIQUID CLEANING	COMPOSITION

following statement is a full description of this invention, including the best method of performing it known

FRAGRANT LIQUID CLEANING COMPOSITIONS

The present invention relates to fragrant liquid cleaning compositions. More particularly it relates to thickened fragrant liquid cleaning compositions with longer lasting olfactory properties.

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Liquid cleaning compositions which contain a perfume are well known in the art. They are formulated and used for a variety of purposes, mainly for general purpose cleaning, such as the cleaning of hard surfaces, e.g. tiles, kitchen sinks, floors, baths, lavatories, etc. For particular purposes, such as bathrooms, lavatories, etc. often a higher level of perfume is included in such cleaning compositions in order to achieve a fragrant atmosphere in such places.

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However, the fragrance thus provided is often only of relatively short lifetime, as the perfume evaporates into the air and is, as it were, diluted by the air. Moreover, often the cleaning composition has only a short residence time on the substrate to be cleaned, thus leaving relatively little time for the perfume to evaporate and impart a pleasant fragrance to the substrate and its surroundings. If the cleaning operation is followed by a rinsing step, whereby the cleaning composition is removed, this causes a further reduction of the residence time of the perfume present on the surface of the substrate.

It is an object of the present invention to increase

the residence time of the fragrant cleaning composition on the surface of the substrate treated therewith and reduce drainag losses by subsequent rinsing.

It is another object of the present invention to reduce the level of perfume in such liquid cleaning compositions

while maintaining or even improving the longevity of the fragranc imparted to the substrate and its surroundings.

5 It has now been found that these and other objects of the invention can be achieved by inclusion of the perfume in a liquid cleaning composition which is shear thinning. Such shear thinning liquid cleaning compositions are obtained according to the present invention by inclusion in an aqueous liquid surface-active agent-containing cleaning composition of a polymer which is shear thinning, thus causing the product containing this polymer to be sufficiently thick when draining but sufficiently thin when being dispensed from the container.

The polymer should have a viscosity, at 25°C in deionised water at a polymer concentration of 3% by weight or less, of between 0.3 and 0.006 Pa.S, preferably between 0.15 and 0.006 Pa.s and particularly preferably between 0.04 and 0.006 Pa.s, at a shear rate of 2000 sec⁻¹, and at a shear rate of 30 sec⁻¹ a viscosity of between 0.7, preferably 0.5 and particularly preferably 0.3 Pa.s and 0.03 Pa.s, and a viscosity at a shear rate of 0.5 sec⁻¹ of between 0.6 Pa.s, preferably 1.1 Pa.s and particularly preferably 2.0 Pa.s and 50 Pa.s.

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The polymer should furthermore be compatible with the surface-active agents present in the cleaning composition. Suitable examples of polymers to be used according to the present invention are biopolymers such as the xanthan gums and derivatives thereof, such as Kelzan S, a partially acetylated xanthan gum ex Kelco Co., Shell-flo-XA ex Shell Chemicals Ltd, Enorflo-XA ex Shell Chemicals, Rhodapol ex Rhôn -Poul nc, cross-linked polyaczylates, such as Carbopol x B.F.Goodrich Co. Ltd, succinoglucans, such as Sh llflo-S ex Sh ll Chemi-

cals Ltd, acrylic copolymers such as E.P. 1996 ex National Adhesives and Resins Ltd.

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The amount of polymer used in the cleaning composition generally ranges from 0.1-3.0%, usually from 0.25-1.0%, and preferably from 0.4-0.8 by weight. The liquid cleaning composition comprises furthermore as essential ingredients one or more detergent active materials which can be anionic, nonionic and zwitterionic type detergent actives or mixtures thereof. Usually anionic synthetic detergents, such as the alkylbenzene sulphonates, alkanesulphonates, alkylsulphates, alkylethersulphates or mixtures thereof can be used. A typical example thereof is a mixture of sodium dodecyl benzenesulphonate and a sodium salt of a sulphated C₁₂-C₁₅ primary linear alcohol condensed with 3 moles of ethylene oxide. In general the amount of active detergent material in the composition ranges from 0.05-20%, usually from 0.1-15% and preferably from 2-10% by weight.

Another essential ingredient of the cleaning composition is a perfume, by which is to be understood either a single fragrant compound or a mixture of various compounds. Any type of perfume may be used. The amount of perfume in the composition may vary from 0.1-1.5%, usually from 0.2-1.0% and preferably from 0.25-0.8%.

Furthermore, other ingredients commonly encountered in such compositions may also optionally be included, such as builders, sequestering agents, dyes, preservatives, bleaches, bleach activators, solvents, enzymes, foam controlling agents, hydrotropes and so on. The liquid medium of the composition usually is an aqueous medium.

The invention will further be illustrated by way of Example.

EXAMPLE 1

The following formulations were prepared.

	5		1 by w	<pre>\$ by weight</pre>	
			λ	В	
		Sodium alkylbenzenesulphonate	1.8	1.8	
		Sodium salt of sulphated C_{12} - C_{15}			
• ••	10	linear primary alcohol condense	ed		
••••		with 3 moles of ethylene oxide	1.8	1.8	
:		Perfume	1.0	1.0	
••••		Industrial Methylated Spirit	4.0	4.0	
•••		Dye	0.007	0.007	
••••	15	Sodium hexametaphosphate	0.5	0.5	
••••		Formalin	0.5	0.5	
•••		Polymer	0.52	0.6	
		Water	ad 100 ad	100	
	20	In formulation A the polymer was Natrosol 250 HBR, a cellulose-based polymer ex Hercules, Inc. and in formulation B the polymer was Kelzan S. The latter polymer had the following viscosity (0.6% in deionised water at			
•	25	25°C): 0.014 Pa.s ($\gamma = 2000 \text{ sec}^{-1}$), 0.4 Pa.s ($\gamma = 30 \text{ sec}^{-1}$) and 8 Pa.s ($\gamma = 0.5 \text{ sec}^{-1}$).			
••••		The concentration of the polymers was chosen to match the dispensing characteristics of both formulations.			
	30	These formulations were compared in the following manner: On to ceramic tiles, 15 g of each product was dosed and the tiles were placed in a vertical position			
		to allow the products to drain. After half an hour, only 6% of formulation A remain d on the tile, whereas			
٠.	35	43% of formulation B wa still pr			

In another xperiment, plastic til s were imm reed in

the test products and then drained vertically for l hour without any further ventilation. The perfume impact was then assessed by a panel using a magnitude estimation (ME) technique. With formulation A, an ME of the perfume strength of 80 was reached, whereas with formulation B the perfume strength was assessed to have an ME of 130.

The results thereof showed that the Kelzan-containing formulation produced a significantly longer lasting perfume intensity.

EXAMPLE 2

The following formulation equally produced a longer lasting perfume intensity, and imparted a glossy appearance to the hard surfaces treated therewith:

		8 by weight
20	Sodium alkylbenzenesulphonate	0.25
	Sodium salt of sulphonated $C_{12}^{-C}C_{15}$	•
	linear primary alcohol condensed with	
	3 moles of ethylene oxide	0.25
	Linear primary C ₉ -C ₁₁ alcohol,	
25	condensed with 5 moles of ethylene oxide	1.0
	Perfüñe	0.7
	Dye	0.01
	Formalin	0.75
	Polymer	1.0
30	Water	balance

The polyme: was an acrylic copolymer, obtainable from National Adhesives and Resins Ltd under the name E.P. 1996. This copolymer has th following viscosities (0.9% in deionised wat r at 25°C):

at $\gamma = 2000$ sec⁻¹ 0.036 Pa.s $\gamma = 30$ s c⁻¹ 0.17 Pa.s $\gamma = 0.5$ s c⁻¹ 1 Pa.s

5 In the above examples, the viscosities were measured with a Haake RV 2 rheoviscometer.

THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:

1. A fragrant aqueous liquid cleaning composition comprising from 0.05-20% by weight of one or more surface-active detergents; from 0.1-1.5% by weight of one or more perfumes, and from 0.1-3% by weight of one or more polymers in an aqueous medium, wherein the polymer is shear thinning, having a viscosity, at 25°C in deionised water at a concentration of 3% by weight or less, of between 0.3 and 0.006 Pa.s at a shear rate of 2000 sec⁻¹, of between 0.7 and 0 03 Pa.s at a shear rate of 30 sec⁻¹, and of between 0.6 and 50 Pa.s at a shear rate of 0.5 sec⁻¹.

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- 2. A composition according to claim 1, wherein the polymer has the following viscosities:

 between 0.15 and 0.006 Pa.s at 2000 sec-1,
 between 0.5 and 0.03 Pa.s at 30 sec-1
 and between 1.1 and 50 Pa.s at 0.5 sec-1.
- 3. A composition according to claim 1, wherein the polymer has the following viscosities:

 between 0.04 and 0.006 Pa.s at 2000 sec-1,
 between 0.3 and 0.03 Pa.s at 30 sec-1
 and between 2.0 and 50 Pa.s at 0.5 sec-1.
- A composition according to claim 1, wherein it contains 0.25-1% by weight of the polymer and 0.25-0.8% of the perfume.
- 5. A composition according to claim 1, wherein the polymer is a manthan gum or a derivative thereof, or a succinoglucan, or a cross-linked polyacrylate, or an acrylic copolym r.

DATED THIS 2ND DAY OF SEPTEMBER, 1985.

UNILEVER PLC
By Its Patent Attorneys:
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Pellows Institute of Patent